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Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. ____ 1.

The first step in most stoichiometry problems is to ____.

- a. add the coefficients of the reagents
- c. convert given quantities to volumes
- b. convert given quantities to moles
- d. convert given quantities to masses

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SECTION 12.1 THE

ARITHMETIC OF

EQUATIONS (pages

353-358) This section

explains how to

calculate the amount

of reactants required

or product formed in a

nonchemical process. It

teaches you how to

interpret chemical

equations in terms of

interacting moles,

representative

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Part 12.11A : First, we
will calculate the
theoretical yield based
on the stoichiometry.

Step 1: List the known
quantities and plan the
problem. Known .

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given: mass of KClO_3
= 40.0 g; molar mass
 KClO_3 = 122.55 g/mol;
molar mass O_2 =
32.00 g/mol; Unknown
. theoretical yield O_2
= ? g

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